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Bioenergetics

Bioenergetics is the field of biochemistry and cell biology which deals with the study of the processes by which cells use, store and release energy. The quantitative study of energy relationships in biological system is called bioenergetics.

A central component of bioenergetics is energy transformation, the conversion of energy from one form to another. The biological energy transformations obey the laws of thermodynamics. Energy is necessary for growth and reproduction. We cannot exhibit

any of characteristics of life without a ready supply of energy. In this chapter we will discuss the most fundamental metabolic processes which are photosynthesis and respiration.

4.1 Photosynthesis

Photosynthesis is the biological process that captures light (solar) energy and converts it into chemical energy (i.e. organic molecules, e.g., glucose). It takes place in plants, algae, cyanobacteria and many bacteria.

Photosynthesis is a "redox" process which links non-living world to the living world. It involves the reduction of carbon dioxide into sugars and the oxidation of water into molecular countries. into molecular oxygen. The overall reactions of photosynthesis can be summarized as follows: follows:

# Tit bits

An exergonic reaction is a spontaneous chemical reaction that releases energy. It is catabolic reaction. An endergonic reaction is an anabolic chemical reaction that consumes energy.

4.1.1 Role of Light in Photosynthesis

Light is a form of energy called Electromagnetic energy or radiation. Solar radiation is consist of photons. Photons (Gk. "Phos"=Light) are separate and distinct packets of energy which come from solar radiation. Photons travel in waves, these waves contain energy. Short waves contain more energy than long wave.

The full range of electromagnetic radiation in the universe is called electromagnetic spectrum while visible light (380-750 nm) is only a small part of the spectrum.

Visible light:

Visible light is the part of the spectrum that the human eye can see which is white light. Photons of visible light have just the right amount of energy to promote electrons to higher electron shell in atoms. Leaves absorb only 1% of total light, which falls on them. rest is reflected or transmitted. The synthesis of ATP from ADP or AMP is called phosphorylation which is endergonic process.

### ADP + Pi + energy ATP

In photosynthetic organisms the energy comes from light thus the process of formation of ATP during photosynthesis is referred to as photophosphorylation. Light falls on green tissues thereby water molecules are broken down (photolysis) into H<sup>+</sup> ions, OH<sup>-</sup> radicals and electrons. The OH<sup>-</sup> radicals are collected and reassembled as water and molecular oxygen, both are released into

# Do you know?

Some carotenoids may protect chlorophyll and human eye from intense light by absorbing and dissipating excessive light energy.

atmosphere. The Hydrogen ions (protons) are pumped across the thylakoid membrane into the lumen. H<sup>+</sup> ions are used to convert NADP to NADPH<sub>2</sub> in photosystem I.

4.1.2 Role of Photosynthetic Pigments

A photosynthetic pigment is a pigment that is present in chloroplasts or photosynthetic bacteria and captures the light energy necessary for photosynthesis. Different pigments absorb light of different wavelengths. The light appear in different colours when passed through a prism.

# Carotenoids:

These are a group of yellow, orange, red or brown pigments that absorb blue, violet and green light. They are associated with the chlorophyll inside the chloroplast or occur alone inside the chloroplast. Carotenoids absorb different wavelengths than chlorophyll, so broaden the spectrum of light that provides energy for photosynthesis. The chlorophyll b and carotenoids together are called accessory pigments because they absorb light and transfer the energy to chlorophyll a which then starts the light reaction.

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Carotenoids in flowers and fruits attract insects, birds and other animals for pollination and seed dispersal respectively. They also protect chlorophyll from oxidation by oxygen produced in photosynthesis.

→ Chlorophyll a Carotenoids  $\longrightarrow$  Chlorophyll b

Chlorophylls are green and main photosynthetic pigments which absorb violet, Chlorophylls: blue, orange and red wavelength's, while green and yellow are least absorbed and are reflected (therefore, leaves look green). There are six types of chlorophylls (a, b, c, d, e, and f) out of these only two types occur in chloroplasts of higher plants, i.e. chlorophyll a and b. Chlorophyll c and d are found only in algae while chlorophyll e and f are found only in bacteria.

# Do you know?

There are two types of carotenoids, i.e. Carotenes and Xanthophylls.

Carotenes (Red to Orange):

Carotenes are hydrocarbons with a general formula of  $C_{40}H_{56}$ . Red colour of tomato and chilli are due to carotenes. The most common carotene is beta-carotene which is converted to vitamin A by animals and human beings.

Xanthophylls are yellow pigments that are oxygen containing derivatives of carotenes. Lutein and zeaxanthin  $(C_{40} H_{56} O_2)$  are the two primary xanthophylls found in green leafy vegetables and other foods like eggs. Yellow colour of leaves in autumn is due to lutein.

The xanthophylls of brown algae is called fucoxanthin (C<sub>40</sub> H<sub>56</sub> O<sub>6</sub>). Both carotenes and xanthophylls are lipid compounds, soluble in organic solvents like other lipids.

Chlorophyll a:

It occurs in all photosynthetic organisms except pigmented bacteria thus termed versal photosynthetic pigment as universal photosynthetic pigment. It is also known as primary photosynthetic pigment because it involves in primary photosynthetic pigment. because it involves in primary reaction during photosynthesis, i.e. convert light energy into chemical energy Molecular 6 into chemical energy. Molecular formula of chlorophyll a is  $(C_{55} H_{72} O_5 N_4 Mg)$ .

Chlorophyll b:

Chlorophyll b occurs in all photosynthetic organisms except brown, red and blue green algae. Porphyrin is derivative of Molecular formula of chlorophyll b is (C<sub>55</sub> H<sub>70</sub>O<sub>6</sub> N<sub>4</sub> Mg).

## Structure of Chlorophyll:

Each chlorophyll molecule has two main parts, one flat square part which absorbs light and hydrophilic head. The other part is long anchoring hydrophobic carbon tail.

The head of chlorophyll is composed of four pyrrole rings (pyrrole is five sided unsaturated nitrogen containing compound) having Mg++ in the center, thus it is Mg<sup>++</sup>porphyrin with two side chains.

# Do you know?

porphin, consists of four pyrrole like rings linked by four CH groups in an alternate double and single bonds. If Mg++ or Fe++ are added to porphin then known as Mg++ porphyrin in chlorophyll or Fe++ porphyrin in Heme and cytochrome.

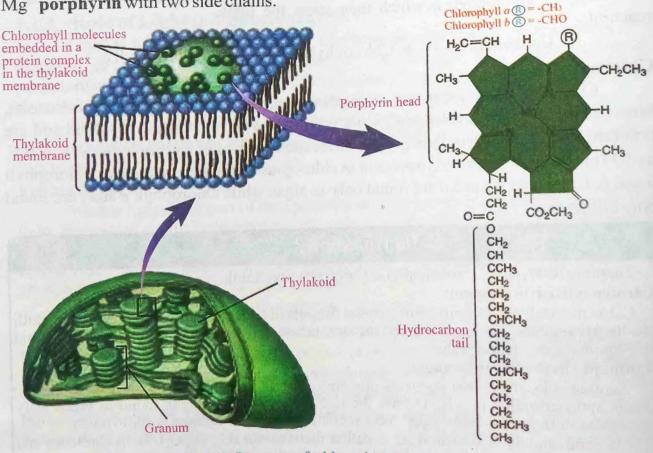


Fig. 4.1 Structure of chloroplast and chlorophyll

a. Acid chain: It is a methyl (CH<sub>3</sub>) ester (H<sub>3</sub>C-O-C=O).

b. Hydrocarbon chain: It is a long hydrocarbon tail which is attached to one of the pyrrole rings and is an alcohol phytol (C<sub>20</sub> H<sub>39</sub>) (it is an ester linkage with propionic acid) (CH<sub>3</sub>-CH<sub>7</sub>-COOH).

Phytol consists of four isoprene units. It is insoluble and serves to anchor the molecule in the membrane of the granum (molecular formula of isoprene is (CH<sub>2</sub>=C-CH<sub>3</sub>-CH=CH<sub>2</sub>)

(C,H.).

Differences Between Chlorophyll a And b:

There is only one difference between chlorophyll a and b that is one of functional group bonded to the porphyrin. In chlorophyll a methyl group (CH<sub>3</sub>) while chlorophyll b aldehyde group (-CHO) is present.

**Role of Pigments in Photosynthesis** 

The clusters of photosynthetic pigments are called photosystem. Each pigment complex is composed of chlorophyll a and b molecules with accessory pigments. When these pigments absorb light they are said to be excited. The light energy is used to boost electrons to a higher energy level which is transferred into chemical energy. The excited state is unstable and molecules will tend to return to its unexcited state.

The energy which is released during this process can be passed from one chlorophyll molecule to another chlorophyll molecule. The instrument which is used to measure relative abilities of different pigments to absorb different wavelength of light is called Spectrophotometer.

## 4.1.3 Absorption Spectrum

It is a measure which exhibits the absorbed amount of the light of different wavelengths (different colours) from the visible spectrum of light. The main photoreceptors are chlorophyll a and chlorophyll b which absorb violet blue (430 nm) and red light (670 nm). The green light (550 nm) is least absorbed. The absorb light between carotenoids 430-470 nm of light spectrum and transfer it to chlorophyll b then to chlorophyll a. The chlorophyll a and b show different absorption spectra as shown figure 4.2 (a). Chlorophyll a shows absorption peaks at about 680 and 700 nm while chlorophyll b absorption peaks range between 450-475 nm.

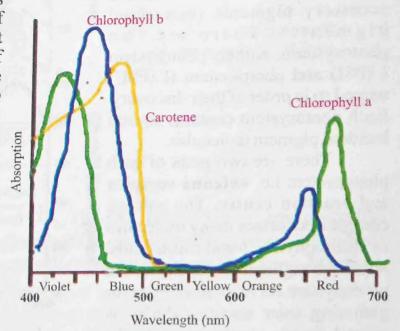


Fig. 4.2 (a) Absorption Spectra

# 4.1.4 Action Spectrum

A graph showing the measure of effectiveness of light of various wavelengths in driving photosynthesis is called action spectrum. Some of absorbed light is released as heat and rest of light is stored in organic compound as chemical energy. Action spectrum of a particular pigment can be calculated by measuring the rate of photosynthesis at each type of wavelength of light. A plant is illuminated with light of different wavelengths. During photosynthesis plant gives off oxygen. As photosynthesis produces oxygen and consumes CO<sub>2</sub>, the rate of production of oxygen or consumption of CO<sub>2</sub> can be used as a measure of the rate of photosynthesis.

# Do you know?

Accessory pigments are photosynthetic pigments that trap light energy and channel it to chlorophyll "a" the primary pigment which initiates the reaction of photosynthesis. Accessory pigmants are carotenoids, phycobilin, some proteins and chlorophyll b, c and d.

4.1.5 Arrangement of Photosynthetic Pigments in the form of Photosystem I and II

Light reaction takes place in the grana of chloroplast. It is initiated when photosynthetic pigments capture light energy. The clusters of photosynthetic pigment complex are composed of chlorophyll a and b molecules and accessory pigments (carotenoid pigments). There are two photosystems, namely photosystem I (PSI) and photosystem II (PSII) named so in order of their discovery. Each photosystem contains several hundred pigment molecules.

There are two parts of each photosystem i.e. antenna complex and reaction center. The antenna complex possesses many molecules of chlorophyll a, b and carotenoids. All these pigment molecules in the photosystem serve as an antenna for gathering solar energy, which is passed from one pigment to the other and finally transferred to the

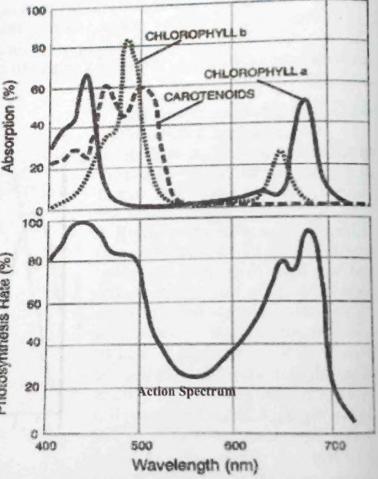
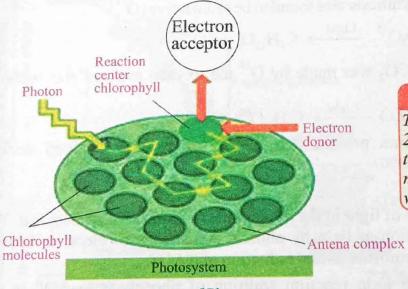


Fig. 4.2 (b) Absorption and Action spectra of different pigments

reaction center. **Reaction center** contains one more molecule of chlorophyll *a* alongwith primary electron acceptor and electron carriers of electron transport system. Electron transport system plays a role in the generation of ATP by **chemiosmosis**. The PSI absorbs light of 700 nm and is called P700 while the PSII absorbs light of P680 nm and is called P680. The **primary electron acceptor** traps the electrons from the reaction center and then passes them on to the series of electron carriers. Electrons have two pathways in the light reaction of photosynthesis; The non-cyclic electron pathway (flow) and cyclic electron pathway. The cyclic is less common and generates only ATP while non-cyclic is predominant and generates both ATP and NADPH<sub>2</sub>.

# Do you know?

The action spectrum is somewhat different from absorption spectrum of chlorophyll. It is more in some wavelengths, such as in 500-600 nm is more than the absorption of green light by chlorophyll. This is because the carotenoids absorb light in this region and pass on some of this absorbed light to chlorophyll, which converts light energy into chemical energy. Similarly when equal intensities of light are given, there is more photosynthesis in red than in blue part of the spectrum.



# Thinking Question

The stomata cover only I to 2% of the leaf surface but they allow proportionally much more gas to diffuse, why?

#### Fig. 4.3 Structure of Photosystem

# 4.1.6 Role of CO<sub>2</sub> in Photosynthesis

Air contains about 0.03 to 0.04 % of CO<sub>2</sub>. This CO<sub>2</sub> is used by terrestrial plants for photosynthesis while aquatic plants use dissolved CO<sub>2</sub> and carbonates present in water as source of carbon. The chloroplasts of guard cells of stomata absorb CO<sub>2</sub>, some of which react with water to form carbonic acid.

$$CO_2 + H_2O \longrightarrow H_2CO_3$$

In the presence of solar energy carbonic acid in the guard cells is decomposed again into water and CO<sub>2</sub>.

Water and carbondioxide are rapidly used in photosynthesis to synthesize organic substances. The entry of CO<sub>2</sub> into the leaves depends upon the opening of stomata.

4.1.7 Role of Water in Photosynthesis

Water is one of the raw materials used in photosynthesis. A film of water present around mesophyll cells of leaf helps to absorb CO<sub>2</sub>. The water molecule is broken down into hydrogen and oxygen by the P<sub>680</sub> during **photolysis**. The hydrogen combines with CO<sub>2</sub> to form organic food and molecular oxygen is released into atmosphere during photolysis of H<sub>2</sub>O. Earlier it was thought that the oxygen released in the process of

### Tit bits

Photolysis is the splitting of a chemical compound by means of light energy i.e., photons e.g., photolysis of water in photosynthesis produces H<sup>+</sup>and O<sub>2</sub>.

photosynthesis comes from  $CO_2$ . Van Neil 1903, was first who observed that water splits during photosynthesis, hydrogen released from water is used to synthesize glucose while  $O_2$  is removed as byproduct. The idea of Van Neil was also supported by another scientist named Hill. In first experiment water was made of  $O^{18}$  and algae were grown in it. The oxygen evolved during photosynthesis was found to be radioactive  $O^{18}$ .

$$6CO_2 + 12H_2O^{18} \xrightarrow{Light} C_6H_{12}O_6 + 6O_2^{18} + 6H_2O$$

In second experiment  $CO_2$  was made by  $O^{18}$  the oxygen evolved was without isotope.

 $6CO_2^{18} + 12H_2O \xrightarrow{\text{Light}} C_6H_{12}O_6^{18} + 6H_2O^{18} + 6O_2$ 

Thus above experiments proved that source of oxygen evolved during photosynthesis was water.

4.1.8 Light Reaction

It occurs in the presence of light in the thyllakoids of granum of chloroplast. The light reaction involves absorption of light by photosystems, flow of electron through electron transport chain i.e. chemiosmosis and reduction of NADP.

**Photophosphorylation:** In light reaction addition of phosphate to ADP in the presence of light is called photophosphorylation. There are two pathways:

(i) Cyclic photophosphorylation (ii) Non Cyclic photophosphorylation

**Cyclic Photophosphorylation:** The cyclic photophosphorylation is less common and generates only ATP while non-cyclic photophosphorylation is predominant and generates both ATP and NADPH<sub>2</sub>.

Non-Cyclic Photophosphorylation: During non-cyclic photophosphorylation electrons move from water through PS-II to PS-I then to NADP.

Photosystem II:

When light strikes the chlorophyll molecules in PSII (p680) its energy causes the

chlorophyll molecule to be activated. The activated chlorophyll loses its two electrons and the positively charged chlorophyll molecule is left in the photosystem with a gap of two electrons. The high energy electrons instead of falling back into the photosystem are captured by primary electrons acceptor of first electron transport chain. The **primary electron acceptor is pheophytin** which then passes the electrons to a plastoquinone (PQ). Now from primary electrons acceptor, the electrons pass along a series of electron acceptor molecules from one to another in oxidation process. These electron acceptors are two **cytochromes** (cytochrome b and f) and **plastocyanin** (PC) (a copper containing protein).

# Production of ATP:

When electrons are passed through electron transport chain, they lose energy. Some of the energy lost by electrons between cytochrome b and cytochrome f is used to make ATP from ADP and Pi. This ATP, which is generated by PS-II will provide energy for Calvin cycle where  $CO_2$  is fixed to synthesize sugar.

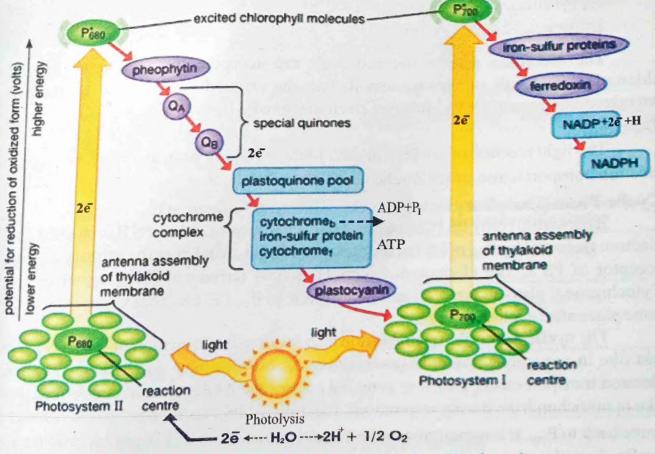


Fig. 4.4 Non-cyclic photophosphorylation (Z-scheme)

### Activity

- 1. Why do we consider the leaves in plants as food factories?
- 2. Trace out the environmental factors that affect the rate of photosynthesis?

#### Photosystem I:

The electrons from PS II pass to PS I. The electrons from plastocyanin are received by another photosystem called photosystem I ( $P_{700}$ ), where these electrons are boosted to high energy state by absorbing a photon of light.

The photoexcited electron of PS I enters in the second electron transport chain. Here electrons are accepted by ferredoxin (FD), which is also an iron containing protein. The enzyme NADP reductase (flavo protein enzyme) by a redox reaction transfers the electrons from ferredoxin to NADP. The NADP combines with electrons and hydrogen to form NADPH<sub>2</sub>.

ATP and NADPH are used in Calvin cycle to produce sugar.

$$NADP^{+} + 2e^{-} + 2H^{+} \longrightarrow NADPH + H^{+}$$

When photosystem II absorbs light water molecule splits into **OH** and **H**<sup>†</sup>. The **OH** ions react to form some water and release oxygen and electrons.

$$4H_2O \longrightarrow 4H^+ + 4(OH^-) + 4e^-$$

$$4(OH^-) \longrightarrow 2H_2O + O_2$$

The electrons from water molecule are accepted by positively charged chlorophyll molecule of photosystem II, thus the emptied hole is filled by the two energized electrons while PS I receives electrons from PS II.

#### **Z-Scheme:**

The light reaction of the PS II and PS I follows zigzag path, therefore, non-cyclic electron transport is also called Z-scheme of light reaction.

### Cyclic Photophosphorylation (Cyclic Electron Transport):

It consists of only PS I and occurs in rare condition i.e. when PS II is blocked. The electrons released by  $P_{700}$  of PS I in the presence of light are taken up by primary electron acceptor of PS I. The electron acceptors consist of ferredoxin (FD), cytochrome b, Cytochrome f, plstocyanin (PC) and finally back to  $P_{700}$  i.e. electrons come back to the same place after cyclic movement.

The cyclic photophosphorylation also result in the formation of ATP molecules just like in non-cyclic photophosphorylation. As the electrons move downhill in the electron transport chain, they lose potential energy and ATP molecules are formed (just like in mitochondrion during respiration). Electrons of PS I do not pass to NADP instead come back to  $P_{700}$ . It is important to note that oxygen and NADPH<sub>2</sub> are not formed during cyclic photophosphorylation.

### Which conditions lead to cyclic electron pathway?

- 1. When production of ATP is low thus Calvin cycle does not begin.
- 2. Due to slow rate of Calvin cycle, NADPH<sub>2</sub> do not oxidize into NADP.

- 3. There are many other enzymatic reactions which use ATP in stroma, thus Calvin cycle becomes slow.
- 4. Limited supply of CO<sub>2</sub> also affects carbohydrate synthesis.

# **Summary of Light Reactions: Requirements:**

- 1. Light
- 2. Enzymes needed for different reactions in the chloroplast
- $H_2O$
- 4. NADP
- 5. ADP and Pi (inorganic phosphate)

### Products:

- 1. Oxygen
- 2. ATP
- 3. NADPH<sub>2</sub>

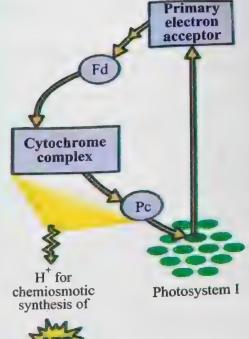




Fig. 4.5 Cyclic photophosphorylation

### Activity

- 1. Draw the Z-scheme for explaining the events of light dependent reactions.
- 2. Draw the labelled structure of chloroplast.

Table 4.1 Comparison between cyclic and non-cyclic photophosphorylation

Non-cyclic photophosphorylation	Cyclic photophosphorylation	
Elections do not come back to the same molecule.	Electrons come back to the same molecule.	
First electron donor is water.	First electron donor is P <sub>700</sub> (PS I).	
Involves both PS I and PS II.	Involves PS I only.	
Last electron acceptor is NADP.	Last electron acceptor is (P <sub>100</sub> ).	
The net products are ATP, NADPH <sub>2</sub> and O <sub>2</sub>	. The Product is ATP only.	

### 4.1.9 Light Independent Reaction or Dark Reaction

The light independent reaction was discovered by Melvin Calvin and coworkers (1950) at the University of California. He was awarded Noble prize in 1961 for his work. Therefore, this cycle is also called **Calvin cycle**. They used radioactive isotope of C<sup>14</sup> in CO<sub>2</sub>. Light independent reactions do not need direct energy of sunlight. It may

occur during day time but it is called dark reaction, so as to differentiate it from the light reaction.

Calvin cycle occurs in the stroma of chloroplast by a scries of reactions in which CO<sub>2</sub> is fixed into carbohydrate (CH,O)n in the absence of light.

The Calvin cycle is completed in three stages:

- Carbon fixation 1)
- ii) Reduction
- Regeneration of ribulose bi-phosphate. iii)

#### Carbon Fixation:

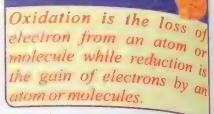
It is first step of dark reaction in which CO2 from air combines with pre-existing five carbon phosphorylating sugar known as ribulose biphosphate (RuBP). As a result an unstable 6-carbon intermediate compound is formed. The enzyme that speeds up this reaction is called RuBP carboxylase; also known as Rubisco.

The six carbon intermediate molecule exists for such a brief time that it cannot be isolated and thus named as an intermediate compound.

# Do you know?

Ruvisco is the most abundani protein on earth.

# Do you know?



### Tit bits

9 ATP and 6 NADPH<sub>2</sub> from light reaction are used in Calvin cycle to produce one PGAL, which can be used to form glucose, fructose etc.

#### Formation of PGA:

The unstable intermediate compound splits into two molecules of three carbon containing phoshoglyceric acid (PGA). It is first identifiable product in dark reaction. Therefore, Calvin cycle is also called C<sub>3</sub> Cycle.

# $CO_2 + RuBP(C_3) \longrightarrow C_6$ (intermediate compound) $\longrightarrow$ 2 PGA(C<sub>3</sub>)

The carbon that was part of CO<sub>2</sub> molecule is now a part of organic molecule. This is called CO, fixation.

# Reduction (Formation of PGAL or G3P):

In this step the product of light reaction that is ATP and NADPH2 are used. Each molecule of phosphoglyceric acid (PGA) receives energy from ATP and H from NADPH<sub>2</sub> forming 3 carbon phosphoglyceraldehyde (PGAL). In this step water is also formed. In reducțion process fixed carbon is reduced into a 3-carbon sugar molecule of PGAL.

# $PGA + ATP + NADPH_2 \longrightarrow PGAL + ADP + Pi + NADP + H_2O$

ADP and Pi and NADP return back to light reaction where ADP is converted into ATP and NADP is reduced into NADPH<sub>2</sub>.

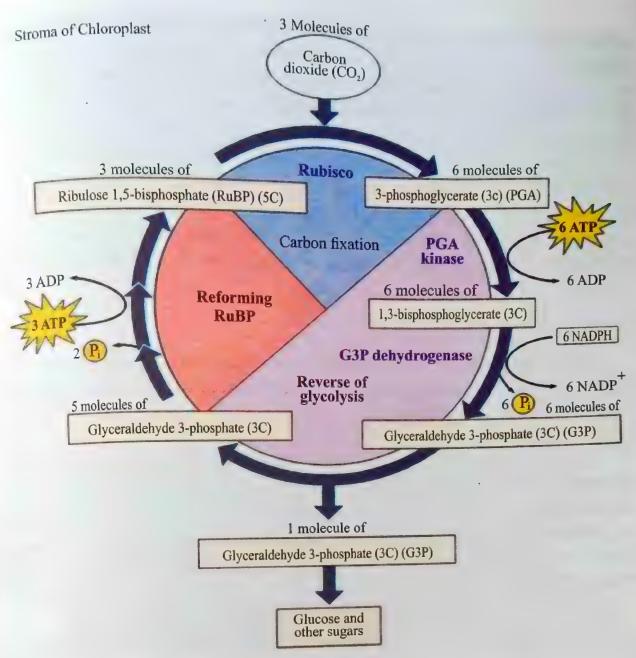


Fig. 4.6 Calvin cycle

# Regeneration of RuBP (Formation of glucose and other organic compounds):

For every turn of Calvin cycle five molecules of PGAL are used to reform three molecules of RuBP, so that cycle can continue. It also uses ATP of light reaction.

$$5 \text{ PGAL} + \text{ATP} \longrightarrow \text{ADP} + \text{Pi} + 3 \text{ RuBP}$$

Thus out of every six molecules of PGAL formed in the reduction stage, only one molecule leaves the cycle, which is to be used by plant for making glucose and other organic compounds.

#### **Use of PGAL**

From PGAL 3C, 4C, 5C, 6C and 7C compounds are produced, all are **105** 

interconvertible. Two PGAL molecules from Calvin cycle are converted into glucose phosphate within chloroplast. Glucose phosphate is then converted to starch. Fixed carbons leave the chloroplast in the form of dihydroxyacetone phosphate (DHAP). It is formed from PGAL. In the cytoplasm DHAP can be used to synthesize the six carbon sugars, glucose and fructose, which are then joined to form sucrose.

Glucose is also used to synthesize cellulose. Glucose is readily converted into amino acids (with the addition of nitrogen). Other compounds like organic acids that is fatty acids and glycerols appear quite rapidly in the cell during photosynthesis. Glucose accumulates more than other compounds, so it was observed more readily by early investigators in chemical analysis. Other compounds can be seen by more sensitive methods.

4.2 Cellular Respiration

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C

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Respiration is a series of complex oxidation reduction reactions in living things. In this process cells get energy through the break down of various organic substances. There are two types of respiration aerobic and anaerobic.

Aerobic Respiration: (Gk. "Aeros" air)

Aerobic respiration takes place in the presence of molecular oxygen. Glucose is a high energy molecule and its breakdown product is CO<sub>2</sub> and H<sub>2</sub>O, which are low energy molecules thus the stored energy is released. The electrons are removed from substrate (e.g., glucose) and eventually received by oxygen atom which combines with H<sup>+</sup>to form water. The overall equation of aerobic respiration for breakdown of glucose can be written as follows:

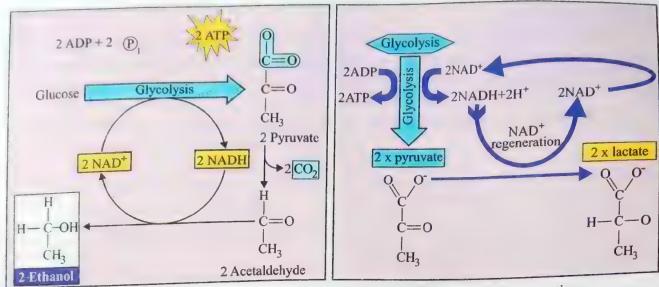
$$C_6H_{12}O_6 + 6O_2 + 36ADP + 36Pi$$
 Enzymes  $6CO_2 + 6H_2O + 36ATP$ 

**Anaerobic Respiration:** 

Anaerobic respiration takes place in the absence of molecular oxygen, it is also known as **fermentation**. It is incomplete oxidation reduction reaction. The energy released from the substrate (glucose) is a result of its molecular rearrangement and some of this energy is available to the cell. The NADH is oxidized to NAD, it is called fermentation because glycolysis is followed by the reduction of pyruvate by NADH to either alcohol and CO<sub>2</sub> or lactate.

#### **Alcoholic Fermentation:**

In primitive cells and cells of some eukaryotic organisms such as yeast and plants, pyruvate is further broken down by alcoholic fermentation into alcohol and  $CO_2$ .



(a) Alcohol Fermentation

Fig. 4.7 Fermentation

(b) Lactic Acid Fermentation

### **Lactic Acid Fermentation:**

It takes place in many bacteria, animals and muscles of human. Each pyruvate molecule is converted into lactic acid in the absence of molecular oxygen.

### Process of cellular respiration:

It takes place in four steps.

- 1. Glycolysis
- 2. Oxidation of pyruvic acid
- 3. Krebs cycle
- 4. Electron transport chain

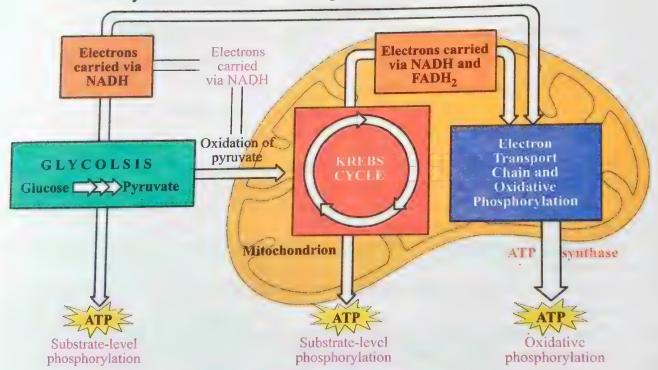


Fig. 4.8 Process of cellular respiration

Glycolysis
Glycolysis takes place in the cytoplasm and it is the break down of glucose into Glycolysis takes place in the cytoplasm and it is the cytoplasm and it is the cytoplasm and evolved before the krebs entry cycle two pyruvate molecules. It is found in all organisms and does not require molecular organisms and does not require molecular organisms. two pyruvate molecules. It is found in all organisms and does not require molecular oxygen and FTC. This occurs in cytosol of cytoplasm and step is catalyzed by an enzyme and FTC. This occurs in cytosol of cytoplasm and does the is catalyzed by an enzyme, thus probably first life was anaerobic bacteria). Each step is catalyzed by an enzyme, thus probably first life was anaerobic bacteria. orobably first life was anaerobic bacteria). Each step of the step of all enzyme, sen orobably first life was anaerobic bacteria). Each step of the st

Hexokinase ATP Mn<sup>2+</sup> 2x1-3, Diphosphoglycerate Glucose 6 phosphate Phosphohexose Isomerase Mg<sup>2+</sup> 2x3, Phosphoglycerate Fructose 6 phosphate Phosphoglycero mutase Phosphoftructo--kinase ADP 2x2, Phosphoglycerate

Mg<sup>2+</sup>

Enolase

H<sub>2</sub>0 Fructose 1, 6 diphosphate 2x Phosphoenol pyruvate Dihydroxyacetone Phosphotroise ----- Glyceraldehyde 3-phosphate 3-phosphate Net 2ATP are produced. 2 x Pvruvate

Fig. 4.9 · Glycolysis

Preparatory Phase (energy investment phase):

In this phase breakdown of glucose occurs and energy is utilized, the steps are:

Phosphorylation of glucose takes place by ATP which produces an activated 1. glucose 6- phosphate molecule.

Glucose 6- phosphate is converted by an enzyme to its isomer fructose 6-

Another ATP molecule transfers a second phosphate group forming 1,6

biphosphate fructose.

The 1, 6 biphosphate fructose splits into two molecules of 3-carbon molecules (Phosphoglyceraldhyde (PGAL) and dihydrooxyacetone phosphate) which are isomers and readily interconvertible.

Oxidative Phase (Energy yielding phase):

Two electrons or two hydrogen atoms are removed from the molecule of PGAL which is oxidized and these electrons are transferred to a molecule of NAD which 1. is reduced. Inorganic phosphate is present in the cell, from which a second phosphate is donated to the molecule forming 1,3 bi or diphosphoglycerate (BPG or DPG).

DPG is converted to 3 phosphoglycerate (3-PGA). Meanwhile a phosphate bond 2.

is transferred from DPG to ADP forming ATP.

3 PGA is converted to 2 phosphoglycerate (2PGA).

From 2 PGA a molecule of water is removed and phosphoenol pyruvate (PEP) is 3. 4.

formed.

PEP then gives up its high energy phosphate which converts ADP to ATP. The product is pyruvate or pyruvic acid (C<sub>3</sub>H<sub>4</sub>O<sub>3</sub>). It is equivalent to half glucose 5. molecule that has been oxidized to the extent of losing two electrons as hydrogen atoms.

# 4.2.3 The Oxidation of Pyruvic Acid

It takes place into two stages.

Oxidation of pyruvic acid to form Acetyl Coenzyme A. 1.

Oxidation of Acetyl Coenzyme A.

Oxidation of Pyruvic Acid:

It is a transition reaction during which CO<sub>2</sub> is released. The oxidation of pyruvic acid is called transition reaction because it connects glycolysis and krebs cycle. In this reaction pyruvate is converted to 2-carbon acetyl Co A by attaching coenzyme A. It gives off carbon dioxide. This is an oxidation reaction in which electrons are removed from pyruvate by dehydrogenase that uses NAD as a coenzyme. This reaction occurs twice for each original glucose molecule.

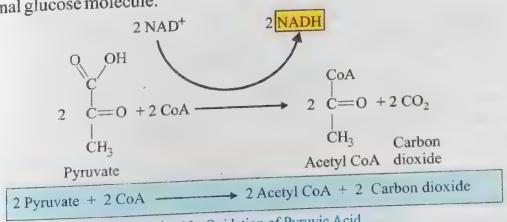


Fig. 4.10 Oxidation of Pyruvic Acid

Oxidation of Acetyl Co enzyme A:

It takes place through krebs cycle. As a first step 4-C compound oxaloacetate binds with 2-carbon acetyl CoA to become 6-carbon compound. This 6-carbon compound passes through a series of electron yielding oxidation reactions. Two carbon dioxide males dioxide molecules are given off. Finally regenerating 4-carbon compound which is free

to bind another acetyl Co A. This cycle is called citric acid cycle or krebs cycle.

4.2.4 Citric Acid Cycle or Krebs Cycle

This is cyclic metabolic pathway located in the matrix of mitochondria. The krebs cycle was named after Sir Hans krebs a British scientist who discovered it in 1930.

Do vou know?

Krebs cycle is also called Tricarboxylic acid cycle because each of its first three reaction has three molecules of carboxylic acid.

Steps of the Krebs cycle:

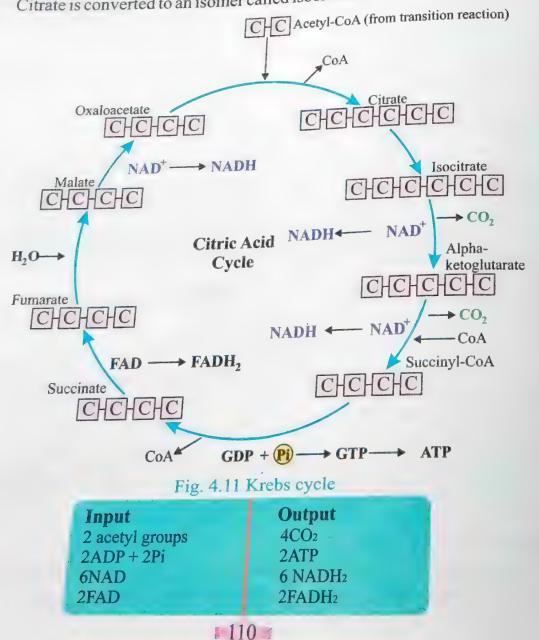
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b

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(

- At the start of this cycle the (2-C) acetyle group (produced by transition reaction) joins with a (4-C) oxaloacetate molecule, forming 6-carbon citrate molecule.
- Citrate is converted to an isomer called isocitrate. 2.



Isocitrate is oxidized by NAD to 5-C alpha-ketoglutarate, NAD is reduced into 3. NADH<sub>2</sub> and CO<sub>2</sub> is released.

Alpha-ketoglutarate is converted into 4-carbon succinyl CoA and NAD is reduced 4. to NADH<sub>2</sub>, another molecule of carbon dioxide is removed.

4-carbon Succinyl CoA is oxidized to 4-C molecule, Succinate. GTP is formed 5. which reacts with ADP to form ATP.

- Now Succinate is converted to 4-carbon Fumarate and FAD is reduced into 6. FADH<sub>2</sub>.
- Fumarate combines with water to produce 4 -carbon Malate. 7.

Malate is oxidized by NAD to Oxaloacetate and NADH $_2$  is formed. 8.

Oxaloacetate is again ready to combine with Acetyl CoA to start a new citric acid 9. cycle.

# 4.2.5 Electron Transport Chain (ETC)

The ETC is located in cristae of mitochondria. It consists of series of carriers that pass electrons from one to the other. The electrons that enter the ETC are carried by NADH and FADH<sub>2</sub> formed during krebs cycle and glycolysis.

Whenever hydrogen is removed from a substrate there are seven intermediate hydrogen acceptors to catch the atom. They are NADH reductase complex (FMN and Fe-S), FADH reductase or co-enzyme Q or Ubiquinone (UQ) and four cytochromes that is b, c, a and a, (cytochromes become pink in color when they are reduced. They are protein plus pigment molecules containing iron. They have ability to gain or lose electron. While abiguinone is not protein it is lipid soluble and water insoluble). Electrons are passed to ubiquinone, at this step an electron is split off the hydrogen atom. The proton becomes free and electron is passed successively from coenzyme Q to cytochrome b, c, a and a<sub>3</sub>.

Thinking Questions

Each NADH +  $H^{\dagger}$  gives 3 ATP in electron transport chain, while each FADH, gives 2ATP. Can you guess whv?

### **Steps of Electron Transport Chain:**

The substances in the chain event are alternately oxidized and reduced. 1.

Oxidation is accomplished by the loss of hydrogen in case of NAD, FAD and the 2. coenzyme while oxidation is accomplished by loss of electrons from cytochrome b, c, a and  $a_3$ .

Since two hydrogen atoms are released at a time and cytochrome b through a, can 3. accept only one electron at a time so there are two cytochrome systems to capture

# Do you know?

In krebs cycle the extracted electrons are temporarily housed within NADH and FADH, molecules. These enter in electron transport system where H are removed, ATP and H.O are formed.

after the final transfer from An electron and proton are brought together cytochrome a3. It produces hydrogen. 4.

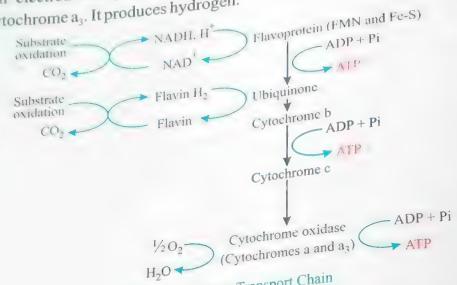


Fig. 4.12 Electron Transport Chain

Molecular oxygen is the hydrogen acceptor and water is the final product. 5.

the hydrogen acceptor
$$O_2 + 4H^{\dagger} + 4e^{-} \longrightarrow 2H_2O$$
to in to ubigu

- Energy is released at three steps, flavoprotein to ubiquinone cyt. b to c, a to a<sub>3</sub>. The released energy is captured by ADP to form ATP. 6.
- Electron transport chain is the main producer of ATP. 7.

# Chemiosmosis

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Chemiosmosis is the synthesis of ATP from ADP 4.2.6 and Pi in the electron transport chain through the joint event of chemical and osmotic processes. The chemiosmotic theory was proposed by Peter Mitchell who got Nobel prize in 1978 for his chemiosmotic theory of ATP production in mitochondria and chloroplasts.

The chemiosmosis can also be defined as the coupling reaction in which synthesis of ATP molecule occurs during the movement of H across a proton gradient. Chemiosmosis generates more ATP as compared to substrate level ATP phosphorylation.

## Tit bits

Chemiosmosis is the movement of ions across a semipermeable membrane, down their electrochemical gradient. For example, generation of ATP by the movement of H across a membrane during cellular respiration or photosynthesis.

# Mechanism of Chemiosmsis:

The mitochordrial membranes have transmembrane channels. These channels can pump protons. The flow of electron induce a change in the shape of protein, thus proton move out of the inner compartment of mitochondria. As a result the proton (H<sup>\*</sup> conc.) in the outer compartment of the mitochondrion becomes greater than that of inner compartment.

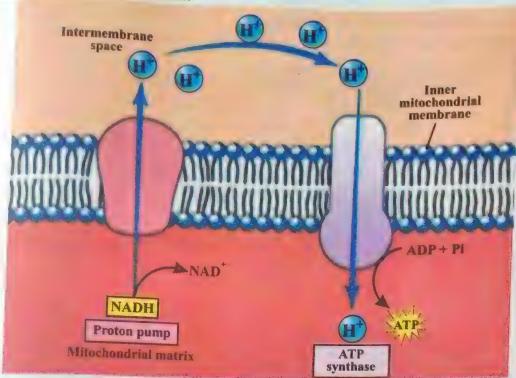


Fig. 4.13 Chemiosmosis

2. Electrical-chemical proton gradient is established between outer and inner membrane. This gradient drives the outer proton across the membrane. Thus the proton move down this gradient between the inner and outer mitochondrial compartments. Their movement induce the formation of ATP from ADP and inorganic phosphate. This process is controlled by an enzyme ATP synthase.

3. The electrons are obtained from the chemical bonds of food molecules in all organisms. This electron removing process needs free oxygen, so it is called aerobic respiration.

### Activity

1. Make a list of differences between photosynthesis, respiration and photorespiration. 2. Draw different steps of ETC.

### 4.2.7 Substrate level phosphorylation

Substrate level phosphorylation occurs in the cytoplasm of the cell during glycolysis and in mitochondrion during the krebs cycle under both aerobic and anaerobic conditions.

ATP formation from ADP and Pi needs input of energy. The energy comes from breakdown of organic molecules in the cells. This type of reaction which releases energy

is called **exergonic reaction**. An enzyme transfers a phosphate group to ADP from a substrate, so ATP molecule is formed. The energy from exergonic reaction is greater than the energy input necessary to drive ATP synthesis. The substrate level phosphorylation appeared very early in the history of organisms. It is recorded in all organisms because initially organisms used carbohydrate as an energy source. Moreover first organisms were anaerobic.

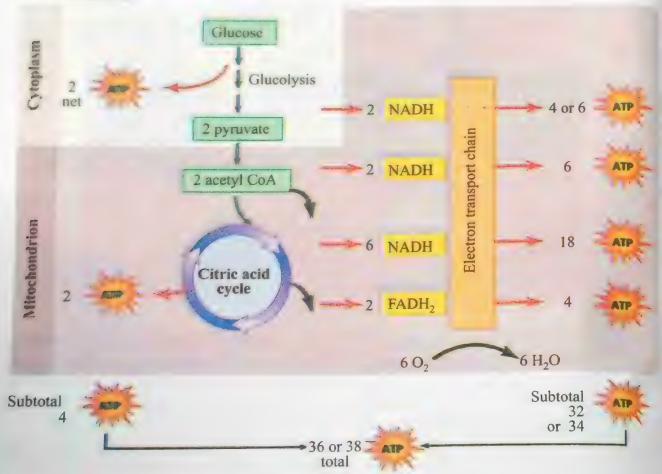


Fig. 4.14 Energy yield per glucose molecule

4.2.8 Importance of PGAL

The preparatory phase of glycolysis completes with the splitting of fructose biphosphate into PGAL and Dihydroxyacetone phosphate (DHAP) and both are interconvertible.

The oxidation of glyceraldehydes 3-phosphate produces 1,3 biphosphoglycerate

and 2 NADH molecules which lead to the formation of pyruvic acid.

PGAL is also formed during the Calvin cycle of photosynthesis, one PGAL molecule leaves Calvin cycle. It is converted into glucose phosphate within chloroplast which is converted into starch.

Fixed carbons leave the chloroplast in the form of dihydroxyacetone phosphate. It is formed from PGAL. The DHAP can be used to make the six carbon sugars, glucose and fructose which become a disaccharide, called sucrose. Now sucrose is transported to other parts of the plants.

4.2.9 Cellular Respiration of Proteins and Fats

Animals and humans besides glucose also consume fats and proteins to harvest energy. Fats are broken down into glycerol and three fatty acids. First the glycerol is phosphorylated then enters the glycolytic pathway at the level of glyceraldehyde 3-Phosphate (PGAL) while fatty acids (2-C)<sub>n</sub> enter in the mitochondrion where their carbons are removed. They form acetyl CoA (2-C) which is entry point for krebs cycle (an 18-carbon fatty acid results in nine acetyl CoA molecules). One gram fat provides about 2.5 times more energy than carbohydrates proteins.

Animals digest proteins into amino acids, if it is in excessive quantity or body is starved then amino acids can be used as fuel. The size of R-group determines whether the carbon chain is oxidizing in glycolysis (Pyruvate) or in the Krebs cycle or cetyl CoA.

Amino acids are degraded, the amine group is removed to yield ammonia this process is called deamination reaction.

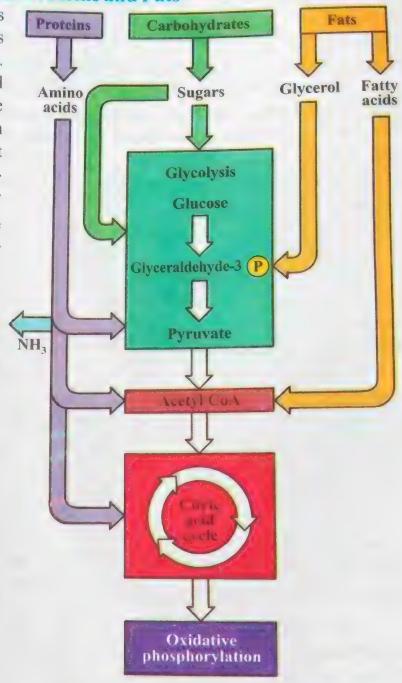


Fig. 4.15 Cellular Respiration of Proteins and Fats

# 4.3 Photorespiration

This process occurs only in photosynthesizing cells of the plants. It is opposite to photosynthesis, because in it oxygen is used instead of CO2 and instead of oxygen, carbon dioxide is released (like respiration). It differs from ordinary respiration of cell which occurs in mitochondria at night and in non-green tissues of plant while photorespiration takes place in the presence of light and only in photosynthetic cell. The oxygen is absorbed but unlike respiration do not produce energy (ATP).

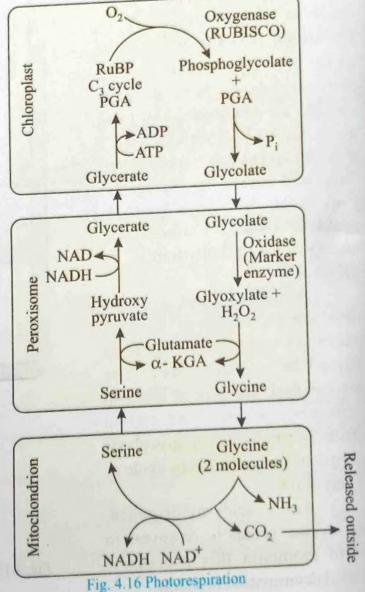
# 4.3.1 How RuBP reacts with oxygen in photorespiration?

Photorespiration is related to the functioning of the enzyme ribulose biphosphate (RuBP) carboxylase which also acts as oxygenase (combines with O2 instead of CO2). The RuBP carboxylase is also known as Rubisco. When rubisco acts as carboxylase it

adds CO2 to RuBP (an acceptor molecule) to produce two molecules of PGA while during oxygenase, it adds oxygen to RuBP and produces one molecule of PGA and one phosphoglycolate. The phosphoglycolate loses its phosphate to become glycolate. There are some algae which can excrete glycolate but higher plants cannot excrete it. Therefore, plants must convert it back to intermediate in the Calvin cycle. The conversion of glycolate into glycine amino acid takes place by a series of reactions in mitochondria, chloroplast and other cellular parts especially in peroxisomes.

Glycolate --- glycine amino acid

Glycine diffuses into mitochondria where every two glycine molecules are converted into serine amino acid and CO2. Serine + CO<sub>2</sub> 2 glycine -



This entire pathway is called photorespiration in which RuBP is converted into serine and CO<sub>2</sub> which uses ATP and NADPH<sub>2</sub> produced during light reaction like Calvin cycle.

4.3.2 Disadvantages of photorespiration

It is reverse to Calvin cycle (here CO<sub>2</sub> is released instead of being fixed into carbohydrates). Photorespiration reduces the amount of carbon fixation into carbohydrates by 25%. The role of photorespiration in plants is not thoroughly understood. It is presumed that photorespiration may be necessary for the assimilation of nitrates from the soil.

4.3.3 Photosynthesis in C4 plants

In normal process of photosynthesis a 3-C compound called PGA is formed as a first detectable product of photosynthesis. Therefore, these plants are called C<sub>3</sub> plants. However, there are some plants growing in dry and hot conditions which produce a four carbon compound (C4) called oxaloacetate as the first product of photosynthesis in dark reaction. These plants are called C4 plants and this type of photosynthesis is called C4 photosynthesis.

C3 plants use rubisco to react  $CO_2$  with RuBP, on the other hand C4 plants use a different enzyme called phosphoenol pyruvate carboxylase (PEPCO) to fix  $CO_2$  to a compound known as phosphoenol pyruvate (PEP). The PEP is reduced into another molecule called malate. The malate carry  $CO_2$  to a special type of cells called bundle sheath cells where Calvin cycle proceeds.

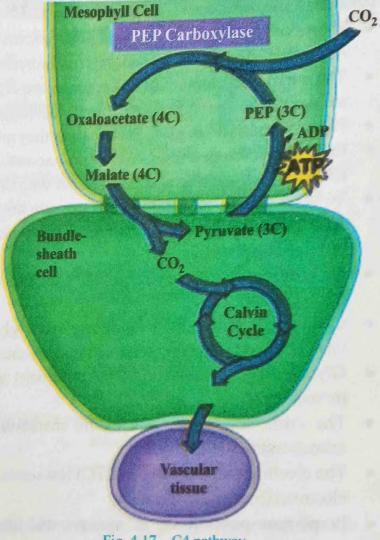


Fig. 4.17 C4 pathway

Table 4.2 Comparison between C<sub>3</sub> and C<sub>4</sub> Plants

### C<sub>3</sub> Plants

# C<sub>4</sub> Plants

In C<sub>3</sub> chloroplasts are located only in mesophyll cells of leaf.

In these plants all the mesophyll cells carry out Calvin cycle.

In high temperature photosynthesis is low.

Example: Most of the plants are C<sub>3</sub> plants such as pea, wheat, rice and all woody trees.

In C<sub>4</sub> chloroplasts are present both in mesophyll cells as well as in bundle sheath cells.

In these plants only mesophyll cells fix CO<sub>2</sub> by

In these plants only mesophyll cells fix CO<sub>2</sub> by using PEPCO while the bundle sheath cells carry out Calvin cycle.

In high temperature the rate of photosynthesis is high.

Example: They are found only in angiosperms such as sugar cane, maize and mostly grasses.

## **SUMMARY**

- Photosynthesis is the only biological process that captures energy of sunlight and converts it into organic compounds (carbohydrates).
- The internal membranes of chloroplasts are organized into sac-like thylakoids which are stacked on one another in columns called grana.
- Photosynthesis takes place in two steps: they are light reaction and dark reaction.
- Each photosystem consists of a light-harvesting complex and a core complex. Each core complex contains a reaction center with the pigment (either P700 or P680).
- To build organic molecules, cells use raw materials provided by the light reactions.
   ATP provided by cyclic and noncyclic photophosphorylation while NADPH<sub>2</sub> provided by photosystem I.
- Cellular respiration is the process in which cells acquire energy by breaking down the organic compounds.
- Cellular respiration involves four phases: glycolysis, the preparatory reaction, the citric acid cycle, and the electron transport chain.
- Glycolysis is the breakdown of (6-carbon) glucose into two (3-carbon) pyruvate molecules.
- The citric acid cycle is a cyclic metabolic pathway located in the matrix of mitochondria.
- The electron transport chain (FTC) is a series of electron comics melecules that ross

## EXERCISE

# Section 1: Objective Questions

### **Multiple Choice Questions**

	the best			
Ch	Which among the fallered	om the fol	ons is favourable for cyclic photo-	
1.	phosphorylation:	ng conditi	ons is lavourable to	
			ence of strange balls	
	(b) Aerobic and low light in			
	(c) Aerobic and optimum 1	_		
	(d) Anaerobic and low light			
2.	During the dark reaction of photosynthesis:  (a) Water is split off			
	(b) CO <sub>2</sub> is reduced to organic compounds			
	(c) Chlorophyll is activated (d) Glucose is broken down			
	(d) Glucose is broken down The enzyme that fixes atmos		in C4 plants is:	
3.	. DED 1 1	spiretic CO	Rubisco	
	2 22 1 1	(d)	Hydrogenase	
,	(c) RuBP carboxylase	e in RuRP	which accepts CO <sub>2</sub> are in C3 plants	
4.		S III Rubi		
	is:	(b)	3	
	(a) 2	(d)		
_	(c) 5 Chlorophyll a differs from cl			
5.	CITIO	(b)	-COOH group	
	(a) -CHO group (c) -CH <sub>3</sub> group	(d)	NITT	
	(c) -CH <sub>3</sub> group NADP is:			
6.		(b)	ApartofrRNA	
	(a) An enzyme	(d)	ApartoftRNA	
7	(c) A coenzyme  The compound that enters the	Krebs cy	cle from glycolysis is:	
7.	Citric acid	(b)		
	(a) Citric acid		Acetyl coenzyme A	
	(c) Pyruvic acid			
Filli	n the blanks.	ecule du	ring PS II of light reaction is	
1.	Breakdown of water mol	ecuie dui	ing is it of light reaction is	

B.

called